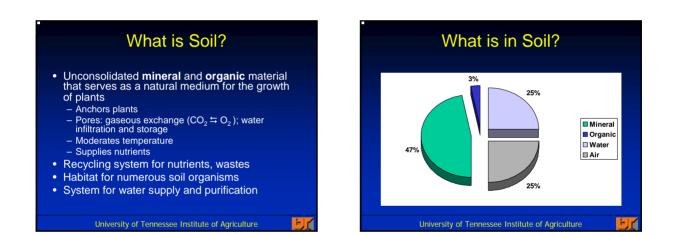


Overview

- What is soil?
- · How is it formed?
- Importance of texture and structure
- What can your color soil tell you?
- Soil pH
- Soil organic matter
- · Essential plant nutrients
- Soil testing
- Tillage

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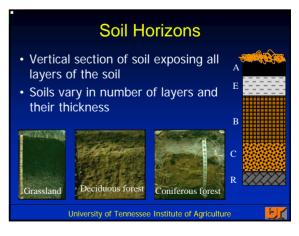


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Soil Forming Factors

- Parent Material
- Geology; wind-blown, glacial or river deposits
- Climate
 - Temperature; rainfall
- Topography / landscape position
 Erosion
- Biota
 - Plants, animals, humans
- Time
 - Weathering; availability of nutrients

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Soil Texture

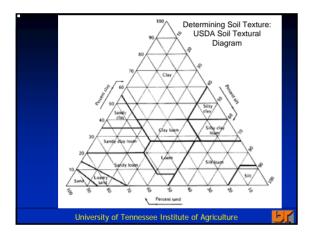
- · Percentage of sand, silt and clay particles
- Does not include material > 2 mm diameter (gravel, rocks etc.), or organic material
- · Permanent characteristic of soil
- Important for many properties: - water holding capacity
 - water movement into and off soil
 - nutrient holding capacity
 - resistance to erosion
 - influences crop management
 - engineering applications

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Soil Texture

- Sand: 0.05 2 mm diameter - single grained, gritty feel, SiO₂ (silicates)
- Silt: 0.002 - 0.05 mm diameter - smooth, floury feel, slightly cohesive, - quartz, feldspars
- Clay: < 0.002 mm diameter
 - stiff, sticky, very cohesive - secondary minerals (weathered)

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Soil Structure • Definition: Aggregation of primary soil separates into compound particles that are separated from adjoining aggregates by planes of weakness • Helps to determine drainage of soil; Influences erosion; Rooting medium; Aeration of subsoil · Five structure types – Granular Platy - Prismatic / columnar - Blocky: angular / sub-angular - "Structureless"

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Soil Color

- Black organic matter
- Grey REDUCED IRON (Fe²⁺), indicates poor drainage, saturated conditions, low oxygen
- Reds, browns, yellows, tans -OXIDIZED IRON (Fe³⁺), good drainage, adequate oxygen
- White high silica content or salts





Soil pH

- Concentration of hydrogen ions
- Measured on pH scale (1 to 14)
- Most crops need pH 5.7 to 6.5
- Micro-nutrient availability less at high pH
- Nutrient uptake; nitrification; acid rain
- Low pH (<5.5) = AI toxicity
 - Correct with calcitic or dolomitic lime
 - Determine with soil test

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Soil Organic Matter

- Source of plant nutrients: N, P, S
- Soil aggregation
- CEC and buffering capacity
- Water holding capacity, air movement, etc.
- Chelation of metals (Zn, Cu)
- C supply for microorganisms
- Surface mulches regulate temperature, moisture

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Humus

- Humic Substances
 - high molecular weight
 - highly aromatic ring structure
 - formed by decomposition and synthesis processes, microbial and chemical
 Very high specific surface area
- Very high Cation Exchange Capacity
- pH dependent
 - 200 to 300 cmol_c/kg
- High water holding capacity
 4 to 5 times its mass

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18 Essential Plant Nutrients Oxygen Hydrogen Carbon Nitrogen Phosphorus Potassium Calcium Magnesium Sulfur Copper Zinc Manganese Molybdenum Iron Chlorine Boron Nickel Cobalt

Role of Nitrogen

- Importance: proteins, chlorophyll, nucleic acids
- Low N = low yields
- Total soil N: < 400 to >8000 kg/ha
 Available soil N = < 5% of total at any time
 - Available Soli N = N_4^+ and NO_3^-
 - only nutrient taken up as cation and anion
 - Remainder = organic forms and $\rm NH_4^+$ "fixed" by clays

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Role of Phosphorus

- The energy currency in the cell
 - ATP -- adenosine triphosphate
 - DNA--deoxyribonucleic acid
 - RNA--ribonucleic acid
- Enhances many aspects of plant physiology
 - photosynthesis, maturation
 - N-fixation, flowering, fruiting, seed production

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- Water relations osmotic regulation
- Energy relations
 - Photosynthesis
 - Translocation
 - Protein synthesis
- Stress resistance

Nutrient Management

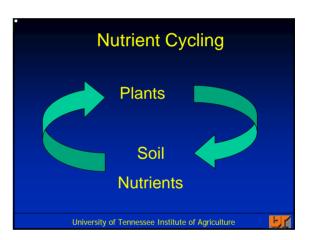
Supply essential nutrients Effective & efficient use of nutrients Minimize environmental degradation Maintain or improve soil quality

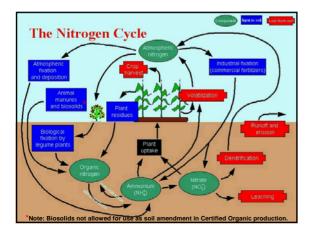
Major Nutrient Cycles

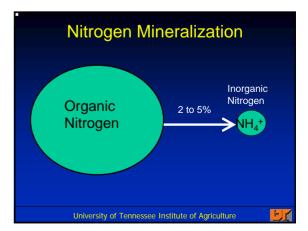
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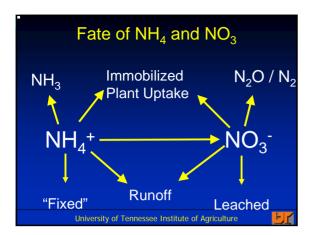
Nitrogen: organic nitrogen, ammonium (NH₄⁺), nitrate (NO₃⁻), ammonia (NH₃)
Phosphorus: phosphorus or phosphate (sold as "P₂O₅"); taken up as PO₄³⁻, HPO₄²⁻, H₂PO₄⁻,
Potassium: potash (sold as "K₂O"); taken up as K⁺

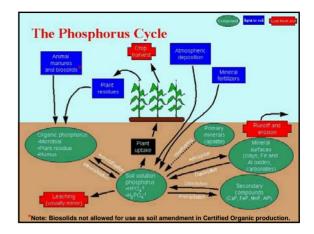
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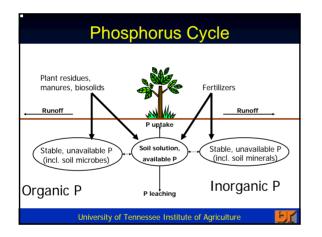








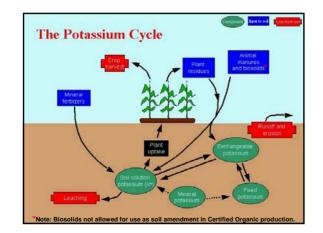




Forms of Phosphorus

Minerals: Ca-P, Fe-P, Al-P, Mn-P Soluble P (depends on pH) $PO_4^{3-} \rightarrow HPO_4^{2-} \rightarrow H_2PO_4^{-} \rightarrow H_3PO_4$ **Fixed Inorganic P** Organic P: available & labile EL

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Transport Mechanisms

Surface Water Runoff	N, P, K
Soil Erosion	N, P, K
Groundwater	N, K
Gas	Ν
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Environmental Concerns

Nitrogen: Air & Water quality Phosphorus: Water quality Potassium: Feed Quality



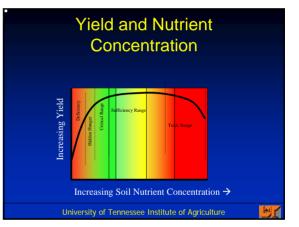
Best Management Practices Reduce Surface Runoff & Erosion Avoid over application Maintain soil pH

Soil Testing

- "Don't guess soil test!"
- Sampling
 - 0 6"
 - Random
- Lab analyses
 - Basic: P, K, pH
 - Cost: \$7
 - http://soilplantandpest.utk.edu/
- Interpretation of results
- Sufficiency vs. Maintenance



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To Till or Not to Till?

- Tillage "a weakness in organic systems?"
- Minimize the negative consequences:
 - Timing of tillage
 - Equipment operation
 - Soil conditions
 - Crop rotation
- Primary (plows) & secondary (harrows) tillage
- No-till 30% residue cover

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Organic Certification and Tillage

- Must document tillage practices as part of Organic System Plans [<u>NOP section</u> §205.201(a)(1)]
- Hand weeding and mechanical cultivation are allowed weed control measures [NOP section §205.206(c)(4)]
- Records must document the frequency of tillage
- Organic inspector will consider whether tillage practices are being used in ways that maintains or improves the physical, chemical, and biological condition of the soil and that minimize soil erosion [NOP section §205.203(a)]

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Tillage

- Weed management
- Destroys soil structure
- Reduces soil organic matter
- Energy intensive
- Increases erosion: 30 % cover = 80 % less erosion

